

Chemistry of Life		Macromolecules				The Cell	
Nonpolar covalent bonds: electrons shared equally between atoms	pH scale: between 0 and 14,	Monosaccharide: monomer of carbs	Lipids: function is energy/protection	Proteins: polymers made of amino acid monomers, linked by peptide bonds	Nucleic acids: DNA and RNA, monomers are nucleotides	Ribosomes: protein factories	Isotonic solution: same amount of solute
Polar covalent bonds: one atom has greater electronegativity unequal sharing of electrons	acids: excess of H ⁺ ions pH < 7					Endoplasmic Reticulum: smooth (synthesis of lipids, package to transport vesicles), me (metabolize carbs, detoxify, storage), Rough (synthesize proteins)	Hypertonic: more solute
Ionic bonds: chemical bonds from attraction between charged ions	Bases: excess of OH ⁻ pH > 7	Polysaccharides: polymers of carbs	Steroids: four rings fused together	Primary structure: amino acid sequence	Made of nitrogenous base, 5 carbon sugar and a phosphate group	Golgi apparatus (receives, sorts, ships)	Hypotonic: less solute
Hydrogen bonds: weak bonds between partial + charged hydrogen atom and electronegative oxygen/nitrogen of another, cause Cohesion (sticking of like molecules), adhesion (sticking of unlike molecules), and transpiration (movement of water molecules in plants)						Endosymbiont theory: mitochondria and chloroplast from prokaryotic cells, have DNA and double membrane	
Specific heat: amount of heat required to raise or lower temp by 1 degree C		Energy-storing polysaccharides: starch, glycogen		Secondary: hydrogen bonding results in alpha helix or beta pleated sheet		peroxisomes: transfer hydrogen to oxygen, detoxify	
Respiration/Fermentation		Structural polysaccharides: cellulose, chitin		Tertiary: complex shapes from bonding between R groups		Energy of Life	
Glycolysis: breakdown two pyruvate (glucose) into 2 pyruvate + 2 H ₂ O, 2 ATP, 2 NADH + 2H ⁺						Catabolic: release of energy by breakdown of complex to simple	
Pyruvate Oxidization: pyruvate turns into CO ₂ , NADH, Acetyl CoA						Anabolic: consume energy to make complicated from simple	
Citric Acid/Krebs cycle: starts w/ acetyl CoA turns into 2CO ₂ , 3NADH, 1ATP, 1FADH						Exergonic: energy released	
ETC: pumping of protons to create a gradient which powers chemiosmosis						Endergonic: requires energy	
Chemiosmosis: ATP synthesis powered by ETC					Quaternary: two or more polypeptide chains into one large protein		
Total yield: 30-32 ATP (2 from glycolysis, 2 from citric acid, 26-28 from oxidative phosphorylation)							
$C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + ATP$							
fermentation: expansion of glycolysis where ATP is produced by substrate level phosphorylation, anaerobic							



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