

Finals

Acids and bases-Definitions

Acids and Bases

Arrhenius Acid: produces H^+ ions in water

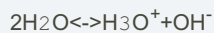
Arrhenius Base: Produced OH^- ions in water

Bronsted Acid: Proton (H^+) donor

Bronsted Base: Proton (H^+) acceptor/rec-eiver

Strong Acids: Donate 100% of the H^+ ion in water Goes to completion

Weak Acids: Donates less than 100% on H^+ ions in water



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