

Finals

Acids and bases-Definitions

Acids and Bases

Arrhenius Acid: produces H^+ ions in water

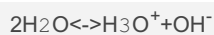
Arrhenius Base: Produced OH^- ions in water

Bronsted Acid: Proton (H^+) donor

Bronsted Base: Proton (H^+) acceptor/rec-eiver

Strong Acids: Donate 100% of the H^+ ion in water Goes to completion

Weak Acids: Donates less than 100% on H^+ ions in water



C

By **danielle_ha_ha**
cheatography.com/danielle-ha-ha/

Not published yet.
Last updated 14th June, 2022.
Page 2 of 2.

Sponsored by **Readable.com**
Measure your website readability!
<https://readable.com>

